

## SYNTHESIS AND INHIBITORY EFFECT OF A TRISUBSTRATE TRANSITION STATE ANALOGUE FOR UDP GLUCURONOSYLTRANSFERASES

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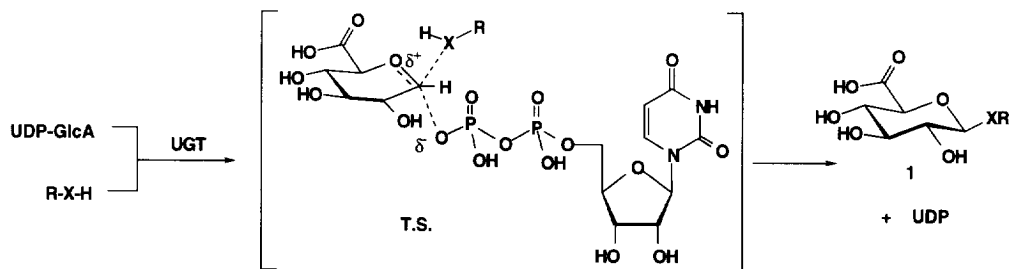
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**Abstract:** Trisubstrate UGT transition state analogue **2** is readily accessible by nucleophilic ring-opening of 1,2-anhydroglucose precursor **5** with diethylmalonate anion followed by reduction of the ethyl ester moieties (**6**→**7**). Subsequent C<sub>6</sub> oxidation (**8**→**9**), NIS/*cat.* TfOH-mediated introduction of the androsteryl/methylene unit (**12**→**15**) and phosphitylation with 5'-uridine phosphoramidite **16** furnished, after oxidation and deprotection, target derivative **2**, the two individual diastereomers of which (**2a** and **2b**) were separated by HPLC. Trisubstrate analogues **2a,b** show a different inhibition pattern for several UGT isoforms, indicating isoenzyme selectivity. Moreover, C<sub>7</sub>-epimers **2a** and **2b** exert a different inhibitory effect on UGT2B15.

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Glucuronidation of xenobiotics and endogenous compounds is a major metabolic detoxification pathway<sup>1</sup> catalyzed by UDP glucuronosyltransferases (UGT's), a family of membrane-bound (iso)enzymes present in mammalian liver cells. The glucuronidation is presumed to proceed *via* the transition state<sup>2</sup> (T.S.) depicted in Figure 1 and comprises transfer of a glucuronosyl residue from α-linked uridine-5'-diphosphate glucuronic acid (UDP-GlcA) to an aglycon (RXH). The resulting β-glucuronide (GlcA-XR, **1**) is readily excreted due to the

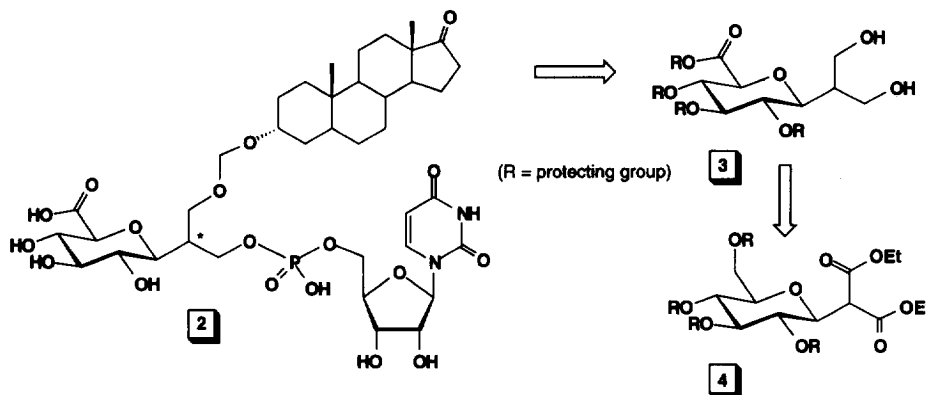
Figure 1



hydrophilicity of the glucuronic acid unit. Enzymatic glucuronidation also plays a pivotal role in the biotransformation of drugs.<sup>3</sup> Consequently, the design<sup>4</sup> of isoenzyme-selective inhibitors<sup>5</sup> of UGT, which may improve the therapeutic efficiency of a drug without affecting the detoxification of endogenous substrates, is a worthwhile goal.

As part of an ongoing program directed towards the design of potential UGT inhibitors<sup>6-8</sup>, we here report the assembly of UGT trisubstrate analogue **2** (see Figure 2), containing a methylene acetal-linked androsterone unit, as well as its inhibitory activity and selectivity towards several UGT isoenzymes.

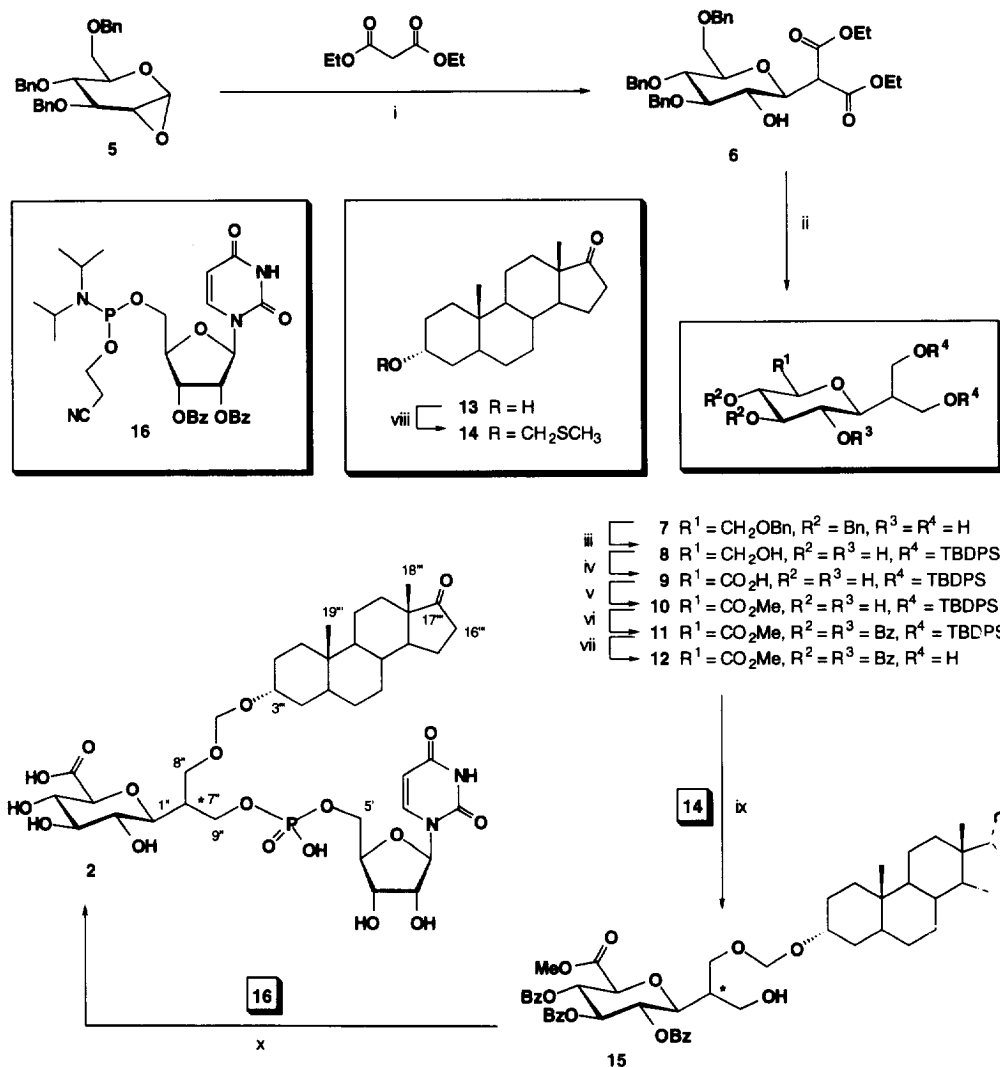
**Figure 2**



Retrosynthetic analysis reveals that target compound **2** is accessible starting from a suitably protected 2-glucuronosyl-1,3-propanediol derivative (**3**), which in turn can be prepared from the corresponding 2-glucosylmalonate (**4**). It was reasoned that C-glucoside **4** can be synthesized by ring-opening of a 1,2-anhydro glucose precursor with malonyl anion.<sup>9</sup> Reaction of known 1,2-anhydro-3,4,6-tri-*O*-benzyl- $\alpha$ -D-glucopyranose<sup>10</sup> (**5**, Scheme 1) with the anion of diethylmalonate did not lead to the desired C-glucoside **6**, but merely resulted in hydrolysis of the 1,2-epoxide into the corresponding 1,2-diol. However, ZnCl<sub>2</sub>-mediated condensation of **5** with diethylmalonate anion<sup>11</sup> gave the 2-glucosyl malonate **6** in a yield of 76%, reduction of which afforded the triol **7**. Silylation of the primary hydroxyl functions in **7** with TBDPSCl, followed by hydrogenolysis of the benzyl protective groups over Pd/C led to the tetraol **8**. The C<sub>6</sub> in **8** was now oxidized using 2,2,6,6-tetramethyl-1-piperidinyloxy free radical (TEMPO) and sodium hypochlorite (NaOCl) under phase-transfer conditions<sup>12</sup> to give carboxylate **9**. Subsequent esterification of the crude acid **9** with diazomethane furnished methyl ester **10** in 62% overall yield from **8**. Benzoylation of **10** and desilylation of the fully protected C-glucuronide **11** with tetra-*n*-butylammonium fluoride (TBAF) in the presence of pyridine-HCl gave the key intermediate diol **12**. The requisite androsterone moiety was now introduced by iodonium ion-mediated reaction of **12** with methylthiomethyl-androsterone **14**. Thus, reaction<sup>13</sup> of commercially available androsterone (**13**) with AcOH/Ac<sub>2</sub>O/DMSO afforded the corresponding methylthiomethyl derivative **14** in 84% yield. *N*-Iodosuccinimide (NIS)/cat. triflic acid (TfOH)-assisted condensation<sup>14</sup> of diol **12** with methylthiomethyl ether **14** led to methylene acetal-linked derivative **15** as a mixture of diastereoisomers (7R:7S = 1:1) in 69% yield. Finally, phosphitylation of the primary alcohol in **15** with the uridine-5'-phosphoramidite **16**<sup>15</sup> under the agency of 1*H*-tetrazole followed by *t*-BuOOH-mediated oxidation of the intermediate phosphite triester, and then removal of the base-labile 2-cyanoethyl and benzoyl protective groups, gave the phosphodiester derivative **2**. Purification of the crude phosphate by HW-40 gel filtration gave homogeneous target compound **2**, the individual C<sub>7</sub>-epimers of which (**2a** and **2b**) were separated by reverse-phase HPLC.

The identity of trisubstrate analogues **2a** and **2b** was unambiguously corroborated by mass spectrometry and  $^{31}\text{P}$ ,  $^1\text{H}$  and  $^{13}\text{C}$  NMR-spectroscopy.<sup>16</sup>

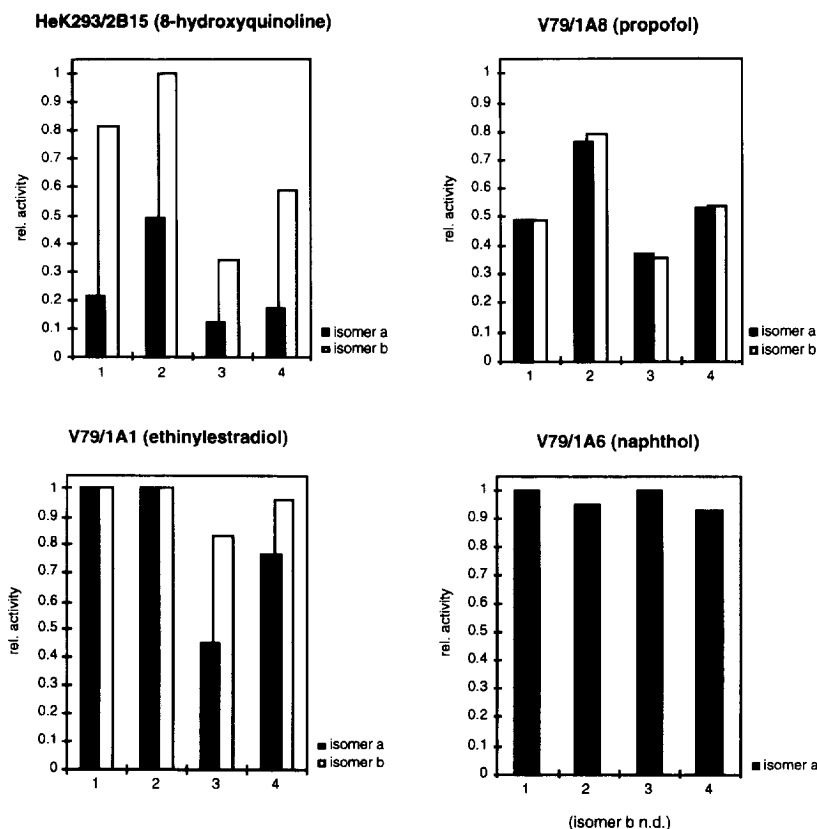
Scheme 1



**Key:** (i)  $\text{NaH}$ ,  $\text{ZnCl}_2$ , THF, 30 min, 76%; (ii)  $\text{LiAlH}_4$ ,  $\text{Et}_2\text{O}$ , reflux, 25 min, 81%; (iii) a. TBDPSCl, DMAP, pyr, 3 h; b.  $\text{H}_2$  (3 atm.), Pd/C,  $t\text{-BuOH}/\text{AcOH}/\text{H}_2\text{O}$  (20:1:2, v/v/v), 12 h, 70%; (iv) TEMPO, NaOCl, KBr,  $\text{NaHCO}_3$ , NaCl,  $(n\text{-Bu})_4\text{NCl}$ ,  $\text{CH}_2\text{Cl}_2/\text{H}_2\text{O}$  (1:1, v/v), 30 min; (v)  $\text{CH}_2\text{N}_2$ ,  $\text{CH}_2\text{Cl}_2/\text{Et}_2\text{O}$ , 5 min (4:1, v/v), 62% (2 steps); (vi)  $\text{BzCl}$ , pyr, 3 h, 91%; (vii) TBAF, pyr-HCl, THF, 12 h, 74%; (viii)  $\text{AcOH}/\text{Ac}_2\text{O}/\text{DMSO}$  (1:4:5, v/v/v), 30 °C, 12 h, 84%; (ix) NIS, cat. TfOH, 1,2-dichloroethane/THF (3:1, v/v), 10 min, 69%; (x) a. 1*H*-tetrazole,  $\text{CH}_2\text{Cl}_2/\text{CH}_3\text{CN}$  (1:2, v/v), 30 min; b.  $t\text{-BuOOH}$  (80% in  $t\text{-BuOO}t\text{-Bu}$ ), 10 min; c. LiOH,  $\text{H}_2\text{O}/\text{MeOH}$  (1:3, v/v), 1 h, 76%.

The inhibitory effect of the two individual diastereoisomers **2a** and **2b** on the UGT activity was investigated using several cloned and expressed UGT isoforms<sup>17</sup> and their appropriate model substrates: V79/1A8 (propofol), HeK293/2B15 (8-hydroxyquinoline), V79/1A1 (ethinylestradiol) and V79/1A6 (naphthol). The relative extent<sup>18</sup> of glucuronidation of the four substrates in the presence of **2** is summarized in Figure 3.

**Figure 3**



**Key:** lane 1: [S] = 50 μM, [2] = 100 μM; lane 2: [S] = 250 μM, [2] = 100 μM; lane 3: [S] = 50 μM, [2] = 500 μM; lane 4: [S] = 250 μM, [2] = 500 μM

The most pronounced inhibitory activity is observed for UGT2B15, the isoenzyme that plays an important role in steroid glucuronidation and the only family 2 isoform used in this study. In this respect, it is of interest to note that isomer **2a** proves to be a better inhibitor than its C<sub>7</sub>-epimer **2b**, which may be attributed to a different spatial alignment of the androsterone and uridine moieties in the respective isomers. On the other hand, both isomers of trisubstrate analogue **2** exert a comparable inhibitory effect on UGT1A8. The degree of UGT1A1 inhibition is considerably lower than that detected for UGT2B15 and UGT1A8. At higher inhibitor

concentration (500  $\mu\text{M}$ ) the two isomers again show a different inhibitory effect. Finally, transition state analogue **2a** showed no inhibition of UGT1A6.

In conclusion, this paper describes an efficient and flexible synthetic route towards the first trisubstrate transition state analogue for UGT. The assembly of inhibitor **2** is based on a novel procedure for the preparation of C-glycosides from 1,2-anhydro sugars (*i.e.* **5**→**6**), regioselective TEMPO-mediated oxidation (*i.e.* **8**→**9**) and iodonium-ion-mediated introduction of the required androsteryl-methylene moiety (*i.e.* **12**→**15**). Trisubstrate analogue **2** shows a different inhibition pattern for several UGT isoforms, indicating isoenzyme selectivity. The most distinct inhibitory effect is observed for UGT2B15, an isoenzyme which plays a pivotal role in steroid glucuronidation. The remarkable difference in UGT2B15-inhibitory activity between diastereomers **2a** and **2b** indicates that the degree of inhibition is influenced by the relative orientation of the aglycon, uridine and glucuronic acid moieties. Inhibitors **2a,b** are quite water-soluble, which implies that their uptake by intact cells, as is required for UGT inhibition *in vivo*, will be relatively slow. However, trisubstrate analogues **2a,b** present promising lead compounds for the future development of more lipid-soluble *in vivo* active UGT inhibitors.

## References and Notes

1. Mulder, G.J.; Coughtrie, M.W.H.; Burchell, B.; *Conjugation Reactions in Drug Metabolism. An Integrated Approach*; Mulder, G.J.; Ed.; Taylor and Francis, London, 51-106 (1990).
2. (a) Noort, D.; Coughtrie, M.W.H.; Burchell, B.; Van der Marel, G.A.; Van Boom, J.H.; Van der Gen, A.; Mulder, G.J.; *Eur. J. Biochem.* **1990**, *188*, 309; (b) Palcic, M.M.; Heerze, L.D.; Srivastava, O.P.; Hindsgaul, O.; *J. Biol. Chem.* **1989**, *264*, 17174.
3. Mulder, G.J.; *Ann. Rev. Pharmacol. Toxicol.* **1992**, *32*, 25.
4. Said, M.; Ziegler, J.-C.; Magdalou, J.; *Quant. Struct.-Act. Relat.* **1996**, *15*, 382.
5. (a) Fournel, S.; Gregoire, B.; Magdalou, J.; Carré, M.C.; Lafaurie, C.; Siest, G.; Caubere, B.; *Biochim. Biophys. Act.* **1986**, *883*, 190; (b) Fournel-Gigleux, S.; Shepherd, S.R.P.; Carré, M.C.; Burchell, B.; Siest, G.; Caubere, P.; *Eur. J. Biochem.* **1989**, *183*, 653; (c) Radomska, A.; Paul, P.; Treat, S.; Towbin, H.; Pratt, C.; Little, J.; Magdalou, J.; Lester, R.; Drake, R.; *Biochim. Biophys. Act.* **1994**, *1205*, 336.
6. Noort, D.; Van der Marel, G.A.; Van der Gen, A.; Mulder, G.J.; Van Boom, J.H.; *Recl. Trav. Chim. Pays-Bas* **1991**, *110*, 53.
7. Noort, D.; Van Straten, N.C.R.; Boons, G.J.P.H.; Van der Marel, G.A.; Bossuyt, X.; Blanckaert, N.; Mulder, G.J.; Van Boom, J.H.; *Bioorg. Med. Chem. Lett.* **1992**, *2*, 583.
8. Said, M.; Noort, D.; Magdalou, J.; Ziegler, J.-C.; Van der Marel, G.A.; Van Boom, J.H.; Mulder, G.J.; Siest, G.; *Biochem. Biophys. Res. Commun.* **1992**, *187*, 140.
9. Hanessian, S.; Pernet, A.G.; *Can. J. Chem.* **1974**, *52*, 1267.
10. Halcomb, R.L.; Danishefsky, S.J.; *J. Am. Chem. Soc.* **1989**, *111*, 6661.
11. In a similar way,  $\text{ZnCl}_2$ -mediated reaction of diethylmalonate anion with 1,2-anhydro-3,4,6-tri-*O*-benzyl- $\alpha$ -D-galactopyranose provided the respective  $\beta$ -C-galactoside (74% yield). It was also established that C-glucoside **6** undergoes a stereoselective cyclization to the corresponding (2*R*)-3-ethyl-2-(3',4',6'-tri-*O*-benzyl- $\beta$ -D-glucopyranosyl)-1,2-malonyl lactone upon prolonged treatment with NaH (3 eq., 12 h, 76% from **5**). Finally, transformation of C-glucoside **6** with DMSO/ $\text{Ac}_2\text{O}$  and subsequent reduction of the intermediate 2'-ulose with  $\text{NaBH}_4$  afforded the corresponding 2-( $\beta$ -D-mannosyl)-malonate (79% yield).
12. Davis, N.J.; Flitsch, S.L.; *Tetrahedron Lett.* **1993**, *34*, 1181.

13. Zavgorodny, S.; Polianski, M.; Besidsky, E.; Kriukov, V.; Sanin, A.; Pokrovskaya, M.; Gurskaya, G.; Lönnberg, H.; Azhayer, A.; *Tetrahedron Lett.* **1991**, *32*, 7593.
14. Veeneman, G.H.; Van der Marel, G.A.; Van den Elst, H.; Van Boom, J.H.; *Tetrahedron* **1991**, *47*, 1547.
15. Phosphoramidite **16** was prepared by phosphitylation of commercially available 2',3'-di-*O*-benzoyl-uridine with 2-cyanoethyl-*N,N*-diisopropylaminochlorophosphine and *N,N*-diisopropylethylamine (DIPEA, CH<sub>2</sub>Cl<sub>2</sub>, 1 h, 94%).
16. Representative spectral data for diastereomer **2a**: <sup>31</sup>P NMR (D<sub>2</sub>O): δ 0.77. <sup>1</sup>H NMR (D<sub>2</sub>O): δ 7.97 (d, 1H, H<sub>6</sub>, *J*<sub>5,6</sub> = 8.1 Hz), 5.96 (d, 1H, H<sub>1</sub>, *J*<sub>1',2'</sub> = 3.5 Hz), 5.94 (d, 1H, H<sub>5</sub>), 4.75 (d, 1H, OCH<sub>2</sub>O, *J*<sub>H,H</sub> = 7.5 Hz), 4.73 (d, 1H, OCH<sub>2</sub>O), 4.32 (t, 1H, H<sub>2</sub>, *J*<sub>2',3'</sub> = 3.5 Hz), 4.31 (t, 1H, H<sub>3</sub>, *J*<sub>3',4'</sub> = 3.6 Hz), 4.24 (m, 1H, H<sub>4</sub>), 4.17 (ddd, 1H, H<sub>5A</sub>, *J*<sub>4',5A</sub> = 2.4 Hz, *J*<sub>5A,5B</sub> = 11.9 Hz, *J*<sub>5A,P</sub> = 4.2 Hz), 4.09 (ddd, 1H, H<sub>5B</sub>, *J*<sub>4',5B</sub> = 2.9 Hz, *J*<sub>5B,P</sub> = 4.9 Hz), 4.01 (m, 2H, H<sub>9A</sub>/H<sub>9B</sub>), 3.89 (m, 1H, H<sub>3''</sub>), 3.81 (dd, 1H, H<sub>8''A</sub>, *J*<sub>7'',8''A</sub> = 4.9 Hz, *J*<sub>8''A,8''B</sub> = 10.4 Hz), 3.66 (dd, 1H, H<sub>8''B</sub>, *J*<sub>7'',8''B</sub> = 7.3 Hz), 3.60 (d, 1H, H<sub>5''</sub>, *J*<sub>4'',5''</sub> = 8.9 Hz), 3.56 (dd, 1H, H<sub>1''</sub>, *J*<sub>1'',2''</sub> = 10.4 Hz, *J*<sub>1'',7''</sub> = 1.2 Hz), 3.50 (dd, 1H, H<sub>3''</sub>, *J*<sub>2'',3''</sub> = 8.2 Hz, *J*<sub>3'',4''</sub> = 9.6 Hz), 3.47 (dd, 1H, H<sub>2''</sub>), 3.44 (dd, 1H, H<sub>4''</sub>), 2.49 (ddd, H<sub>16''A</sub>, *J*<sub>15''A,16''A</sub> = 0.9 Hz, *J*<sub>15''B,16''A</sub> = 8.6 Hz, *J*<sub>16''A,16''B</sub> = 19.8 Hz), 2.44 (m, 1H, H<sub>7''</sub>), 2.15 (ddd, H<sub>16''B</sub>, *J*<sub>15''A,16''B</sub> = 8.0 Hz, *J*<sub>15''B,16''B</sub> = 2.0 Hz), 1.96-0.78 (m, 20H, 2 × H<sub>1''</sub>/2 × H<sub>2''</sub>/2 × H<sub>4''</sub>/H<sub>5''</sub>/2 × H<sub>6''</sub>/2 × H<sub>7''</sub>/H<sub>8''</sub>/H<sub>9''</sub>/2 × H<sub>11''</sub>/2 × H<sub>12''</sub>/H<sub>14''</sub>/2 × H<sub>15''</sub>), 0.88 (s, 3H, 3 × H<sub>18''</sub>), 0.80 (s, 3H, 3 × H<sub>19''</sub>). <sup>13</sup>C {<sup>1</sup>H} NMR\* (D<sub>2</sub>O): δ 218.1 (C<sub>17''</sub>), 177.4 (C<sub>6''</sub>), 156.5 (C<sub>6</sub>), 154.4 (C<sub>2</sub>), 152.0 (C<sub>4</sub>), 103.2 (C<sub>5</sub>), 92.9 (OCH<sub>2</sub>O), 89.2 (C<sub>1</sub>), 84.0 (C<sub>4</sub>, *J*<sub>4',P</sub> = 8.4 Hz), 78.2 (C<sub>2''</sub>), 76.3 (C<sub>1''</sub>), 74.8 (C<sub>2</sub>), 73.2 (C<sub>3''</sub>), 72.9 (C<sub>4''</sub>), 72.0 (C<sub>3''</sub>), 71.1 (C<sub>5''</sub>), 70.5 (C<sub>3</sub>), 66.0 (C<sub>8''</sub>), 65.4 (C<sub>5</sub>, *J*<sub>5',P</sub> = 4.0 Hz), 64.9 (C<sub>9''</sub>, *J*<sub>9',P</sub> = 4.0 Hz), 54.9 (C<sub>9''</sub>), 51.7 (C<sub>14''</sub>), 49.4 (C<sub>13''</sub>), 41.5 (C<sub>7''</sub>, *J*<sub>7',P</sub> = 8.0 Hz), 40.4 (C<sub>5''</sub>), 36.9, 36.2 (C<sub>10''</sub>/C<sub>16''</sub>), 35.4 (C<sub>8''</sub>), 33.2, 33.1, 31.8, 31.0, 28.7, 26.6, 22.2, 20.4 (C<sub>1''</sub>/C<sub>2''</sub>/C<sub>4''</sub>/C<sub>6''</sub>/C<sub>7''</sub>/C<sub>11''</sub>/C<sub>12''</sub>/C<sub>15''</sub>), 14.0 (C<sub>18''</sub>), 11.5 (C<sub>19''</sub>). MS (ESI): *m/z* = 860 (M-H)<sup>+</sup>.
17. Burchell, B.; McGurk, K.; Brierley, C.H.; Clarke, D.J.; *Comprehensive Toxicology*; Sipes, I.G.; Gandolfi, A.J.; McQueen, C.A.; Eds.; Vol. 3, *Biotransformations*; Guengerich, F.P.; Ed.; Elsevier Science, Amsterdam, 401-435 (1997).
18. The UGT activity in the presence of **2a** or **2b** was related to a control experiment carried out in the absence of inhibitor. The control activities (nmol/min/mg) at substrate concentrations of 250 μM for the four UGT isoforms are: HeK293/2B15 = 0.194; V79/1A6 = 0.235; V79/1A1 = 0.168; V79/1A6 = 1.27. The assay conditions were: 100 mM tris/maleate (pH = 7.4), 5 mM MgCl<sub>2</sub> and substrate concentrations from 25 μM to 500 μM. The inhibitor was dissolved in DMSO and a control curve with 5 μL of DMSO per assay was carried out concurrently with the inhibition study. The assays were started by addition of 10 μL of 20 mM UDP-GlcA (containing 0.1 μCi of [<sup>14</sup>C]-UDP-GlcA per 10 μL). Incubations were run for 40 min and then terminated by the addition of 100 μL of methanol (-20 °C). Assays were left on ice until the protein could be removed by centrifugation at 1000 g for 5 min. The supernatant (150 μL) was directly injected onto HPLC, comprising a binary gradient of 0-86% acetonitrile in 0.05 M ammonium acetate developed over 13 min on a spherisorb 5ODS2 column. [<sup>14</sup>C]-Labelled UDP-GlcA and glucuronide detection was executed by a heterogeneous radiochemical method using a 500 μL flow cell packed with silanized cerium-activated glass as scintillant. Two independent experiments were carried out in each assay.

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